

# Synthesis of new cyclolinear permethyloligosilane-siloxanes

N. A. Chernyavskaya, V. I. Aleksinskaya, and A. I. Chernyavskii\*

A. N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences,  
28 ul. Vavilova, 117813 Moscow, Russian Federation.

Fax: +7 (095) 135 5085. E-mail: chern@ineos.ac.ru

$\alpha,\omega$ -Bis(heptamethylcyclotetrasiloxanyloxy)oligodimethylsilanes were synthesized for the first time by heterofunctional condensation of hydroxyheptamethylcyclotetrasiloxane with  $\alpha,\omega$ -dichloropermethyloigosilanes,  $\text{Cl}(\text{Me}_2\text{Si})_n\text{Cl}$  ( $n = 2, 4$ , or  $6$ ). The compounds obtained were characterized by spectroscopic methods.

**Key words:** heterofunctional condensation, hydroxyheptamethylcyclotetrasiloxane,  $\alpha,\omega$ -dichloropermethyloigosilanes; IR, UV,  $^{29}\text{Si}$  NMR, and mass spectra.

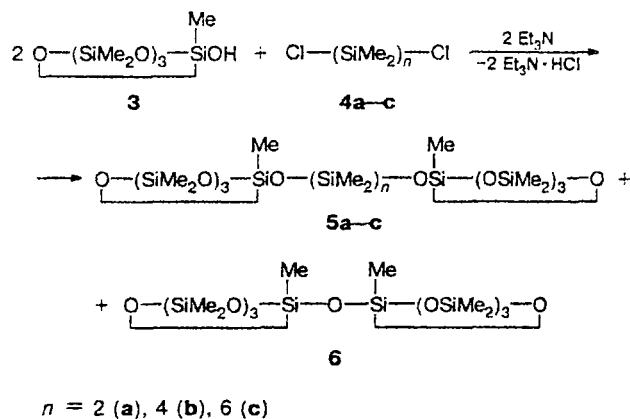
Previously, compounds with the completely methyl environment of the silicon atoms (the so-called "dumb-bell-like" structures<sup>1</sup>) have been prepared, namely,  $\alpha,\omega$ -bis[permethylcyclotri(tetra)siloxy]oligodimethylsiloxanes<sup>2,3</sup> (**1**) and  $\alpha,\omega$ -bis[permethylcyclopenta(hexa)silanyl]oligodimethylsilanes<sup>4–8</sup> (**2**), which contain only the siloxane or silylene units, respectively.

In this work, mixed silane-siloxane compounds with analogous structures, namely,  $\alpha,\omega$ -bis(heptamethylcyclotetrasiloxanyloxy)oligodimethylsilanes (**5**), were first synthesized by heterofunctional condensation of hydroxyheptamethylcyclotetrasiloxane (**3**) with  $\alpha,\omega$ -dichloropermethyloigosilanes  $\text{Cl}(\text{SiMe}_2)_n\text{Cl}$  ( $n = 2, 4$ , or  $6$ ) (**4**) (Scheme 1).

GLC and GLC-mass spectroscopy. It was demonstrated that in all the cases under study, a small amount of bis(heptamethylcyclotetrasiloxanyl) oxide (**6**) (up to 5%) was formed along with oligosilane-siloxanes **5**. In the mass spectrum, compound **6** is characterized by the ion peak  $[\text{M}-\text{Me}]^+$  at  $m/z$  563. In their physical state, oligosilane-siloxanes **5a–c** (Table 1) are intermediate between compounds **1** and **2**, which have purely siloxane and silane structures, respectively. Oligosilane-siloxanes **5a,b** are pale-yellow viscous liquids (more viscous than oligosiloxanes **1**), compound **5c** is a crystalline compound with a low melting temperature, whereas oligosilanes **2** are crystalline compounds with high melting temperatures.

The structures of compounds **5a–c** were confirmed by spectral methods (see the Experimental section).

Scheme 1



The reaction was carried out in ethyl ether at temperatures from 0 to  $-5^\circ\text{C}$  in the presence of triethylamine as an acceptor of HCl. The yields of compounds **5a–c** were 46–64%. The products of the reaction of cyclotetrasiloxane **3** with oligosilanes **4** were analyzed by

## Experimental

The GLC analysis was carried out on an LKhM-8MD chromatograph (a  $0.3 \times 100$  cm stainless steel column; 5% SE-30 on Chromaton N-AW-DMCS; a thermal conductivity detector; the temperature was increased from 30 to  $300^\circ\text{C}$  with a rate of  $12^\circ\text{C min}^{-1}$ ; helium as the carrier gas). The GLC-mass spectrometric analysis was performed on a Kratos-MS-890

Table 1. Yields and characteristics of compounds **5a–c**

Com- po- und	Yield (%)	B.p. / $^\circ\text{C}$ (p/Torr)	$n_D^{20}$	Found Calculated (%)			Molecular formula
				C	H	Si	
<b>5a</b>	64.7	121–122 (0.01)	1.4174	30.79 30.38	7.44 7.65	38.97 39.48	$\text{C}_{18}\text{H}_{54}\text{O}_{10}\text{Si}_{10}$
<b>5b</b>	59.6	158–160 (0.008)	1.4413	32.06 31.92	8.12 8.04	40.21 40.72	$\text{C}_{22}\text{H}_{66}\text{O}_{10}\text{Si}_{12}$
<b>5c</b>	46.8	197–198 (0.007)*	—	33.61 33.08	8.42 8.33	41.17 41.65	$\text{C}_{26}\text{H}_{78}\text{O}_{10}\text{Si}_{14}$

\* M.p.  $37\text{--}38^\circ\text{C}$ .

instrument (a 25 m  $\times$  0.32 mm capillary column; SE-30 liquid phase; helium as the carrier gas; ionizing voltage was 70 eV; the temperature was increased from 30 to 270 °C with a rate of 12 K min<sup>-1</sup>). The <sup>29</sup>Si NMR spectra were recorded on Bruker WP-200 SY and Bruker WP-400 SY spectrometers; the chemical shifts are given relative to Me<sub>4</sub>Si. The IR and UV spectra were obtained on Specord M-80 and Specord M-40 spectrophotometers, respectively.

Cyclosiloxane **3** and dichlorooligosilanes **4a–c** were prepared according to known procedures.<sup>2,9,10</sup>

All reactions were carried out under an atmosphere of dry argon.

**1,2-Bis(heptamethylcyclotetrasiloxanyloxy)tetramethyldisilane (5a).** A solution of dichlorodisilane **4a** (0.63 g, 3.35 mmol) in anhydrous ether (30 mL) was added dropwise to a solution of cyclosiloxane **3** (2 g, 6.7 mmol) and Et<sub>3</sub>N (0.68 g, 6.7 mmol) in anhydrous ether (40 mL) at -5 °C. The reaction mixture was stirred at -5 °C for 2 h and then at -20 °C for 20 h. The Et<sub>3</sub>N  $\cdot$  HCl residue was filtered off. The ethereal solution was washed three times with equal volumes of water and dried with Na<sub>2</sub>SO<sub>4</sub>. After removal of the solvent, the residue was fractionated. Compound **5a** was obtained in a yield of 1.54 g (Table 1). UV,  $\lambda_{\text{max}}/\text{nm}$ : ~199. IR (KBr),  $\nu/\text{cm}^{-1}$ : 2964, 2904 (C–H), 1262, 854, 844, 806 (Si–Me); 1082, 1055 (SiOSi). <sup>29</sup>Si NMR (CCl<sub>4</sub>),  $\delta$ : 1.43 (OSiSiO); -18.94 (OSiOSiOSiO); -19.23 (OSiOSiMe(O–)<sub>2</sub>); -63.74 (MeSi(O–)<sub>3</sub>). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$  (%)): 695 [M–Me]<sup>+</sup> (1.9); [M/2]<sup>+</sup> (77.6).

Mass spectrum of compound **6**,  $m/z$  ( $I_{\text{rel}}$  (%)): 563 [M–Me]<sup>+</sup> (20.8).

**1,4-Bis(heptamethylcyclotetrasiloxanyloxy)octamethyltetrasilane (5b).** Compound **5b** was prepared from cyclosiloxane **3** (3.75 g, 12.6 mmol), Et<sub>3</sub>N (1.27 g, 12.6 mmol), and dichlorooligosilane **4b** (1.91 g, 6.3 mmol) according to a procedure analogous to that described above. The yield was 3.1 g. UV,  $\lambda_{\text{max}}/\text{nm}$ : 236.5. IR (KBr),  $\nu/\text{cm}^{-1}$ : 2958, 2900 (C–H); 1259, 849, 806 (Si–Me); 1071, 1056 (SiOSi). <sup>29</sup>Si NMR (CCl<sub>4</sub>),  $\delta$ : 9.45 (OSiSi<sub>2</sub>SiO); -19.04 (OSiOSiOSiO); -19.46 (OSiOSiMe(O–)<sub>2</sub>); -47.63 (OSiSiSiSiO); -64.20 (MeSi(O–)<sub>3</sub>). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$  (%)): 811 [M–Me]<sup>+</sup> (0.4); 413 [M/2]<sup>+</sup> (15.9).

**1,6-Bis(heptamethylcyclotetrasiloxanyloxy)dodecamethylhexasilane (5c).** Compound **5c** was prepared from cyclosiloxane **3** (2.30 g, 7.7 mmol), Et<sub>3</sub>N (0.78 g, 7.7 mmol), and dichlorooligosilane **4c** (1.62 g, 3.85 mmol) according to a procedure analogous to that used for preparing compound **5a**. The yield was 2.0 g (Table 1). After recrystallization from a 1 : 7 THF–EtOH mixture, compound **5c** was obtained in a yield of 1.7 g. UV,  $\lambda_{\text{max}}/\text{nm}$ : 260. IR (KBr),  $\nu/\text{cm}^{-1}$ : 2962, 2894 (C–H); 1260, 852, 808, 767 (Si–Me); 1079, 1055 (SiOSi). <sup>29</sup>Si NMR (a 9 : 1 CCl<sub>4</sub>–CDCl<sub>3</sub> mixture),  $\delta$ :

9.14 (OSiSi<sub>4</sub>SiO); -19.16 (OSiOSiOSiO); -19.60 (OSiOSiMe(O–)<sub>2</sub>); -40.22 (OSi<sub>2</sub>SiSi<sub>2</sub>O); -45.44 (OSiSiSi<sub>2</sub>SiSiO); -64.35 (MeSi(O–)<sub>3</sub>). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$  (%)): 927 [M–Me]<sup>+</sup> (0.4); 471 [M/2]<sup>+</sup> (67.8).

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